Inverse agonist:

It has full affinity towards the receptor but produces effect completely opposite to that produced by an agonist.

e.g, benzodiazepines (agonist)

B carboline (inverse

agonist)

Antagonist:

Antagonist have only

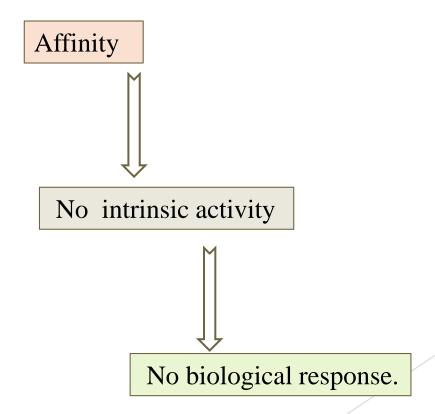
Affinity

No intrinsic activity.

They block the receptor (e.g., naloxone, atropine).

Antagonist:

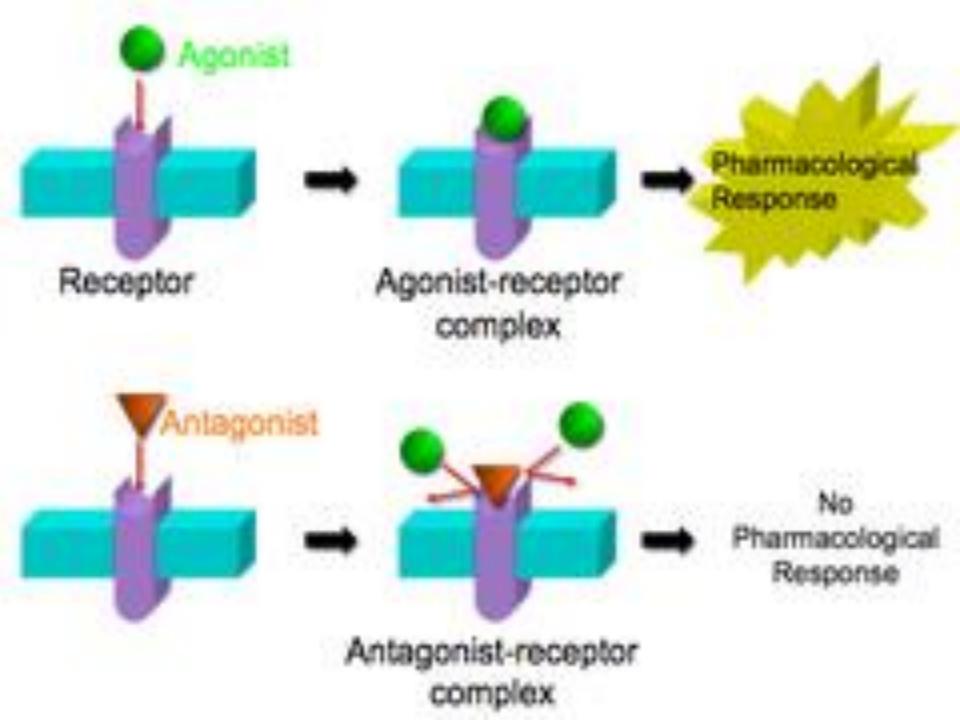
Antagonist have only affinity no intrinsic activity



Types of antagonisms

- Receptor block (pharmacological antagonism)
- a) Competitive antagonism (reversible)
- b) Non-competitive antagonism (irreversible)
- Chemical antagonism
- Physical antagonism
- Physiological antagonism
- Pharmacokinetic antagonism(dispositional)

Pharmacological antagonism



Competitive - antagonism (1)

In this type of antagonism the drug selectively binds to the receptor without activating it but in turn inhibits the binding of an agonist to the receptor.

Surrmountable Block

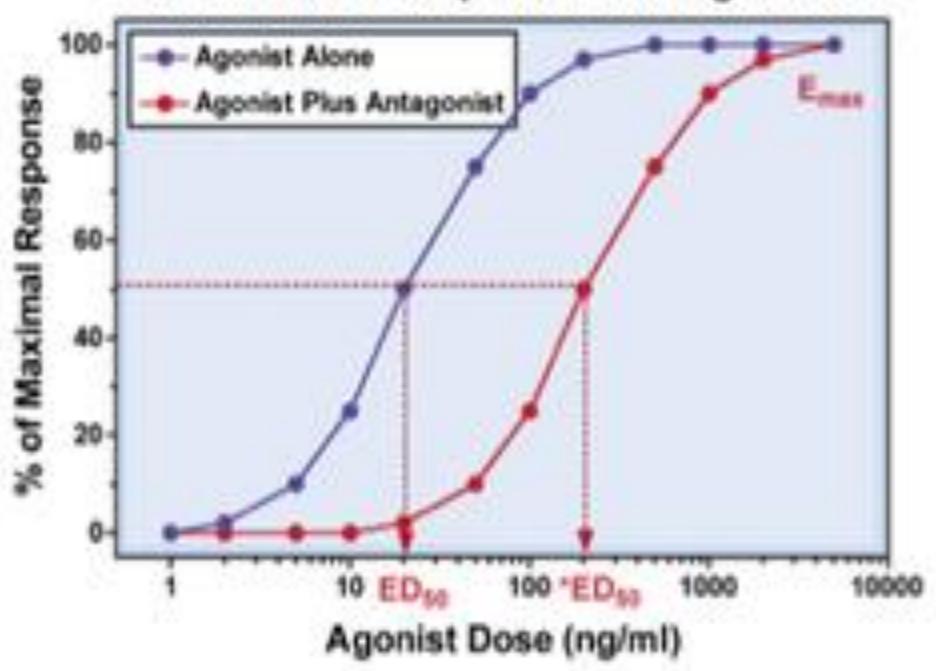
(Weak hydrostatic bonds)

Acetylcholine → MUSCRINIC receptor ← Atropine

Morphine → Opioid receptor ← Nalaxone

(agonists) (antagonists)

Reversible Competitive Antagonism



non-Competitive - antagonism

The antagonist binds to the receptor with a high affinity so that the agonist cannot displace it from the binding site.

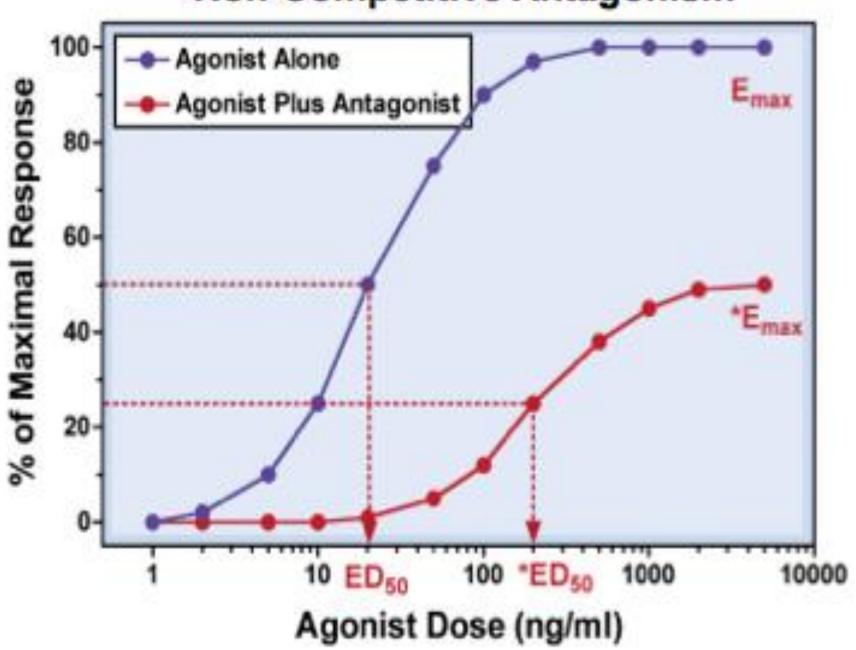
```
(covalent bonds)
```

example

Adrenaline alpha receptor Phenoxybenzamine

(Agonist) (receptor) (Antagonist)

Non-Competitive Antagonism



CHEMICAL ANTAGONISM

In this type, the agonist and the antagonist are combined in a solution and the effect of the agonist is lost.

or

The opposing effect of the two drugs is due to their chemical property.

example

Dimercaprol (chelating agent) form insoluble complexes with metals in metal poisoning

Physiological - Antagonism

In this type of antagonism, a drug produces an effect opposite to that produced by another drug by acting on different receptor or on the same receptor.

example

Adrenaline and Histamine in Anaphylactic shock.

Pharmacokinetic antagonism

In this type of antagonism, any drug altering the absorption, distribution, metabolism or excretion of another drug can alter the concentration of the drug at its receptor site.

example

NaHCO3 increases the excretion of aspirin and thus decreases its concentration.

PHYSICAL ANTAGONISM

The opposing action of two drugs is due to their physical property.

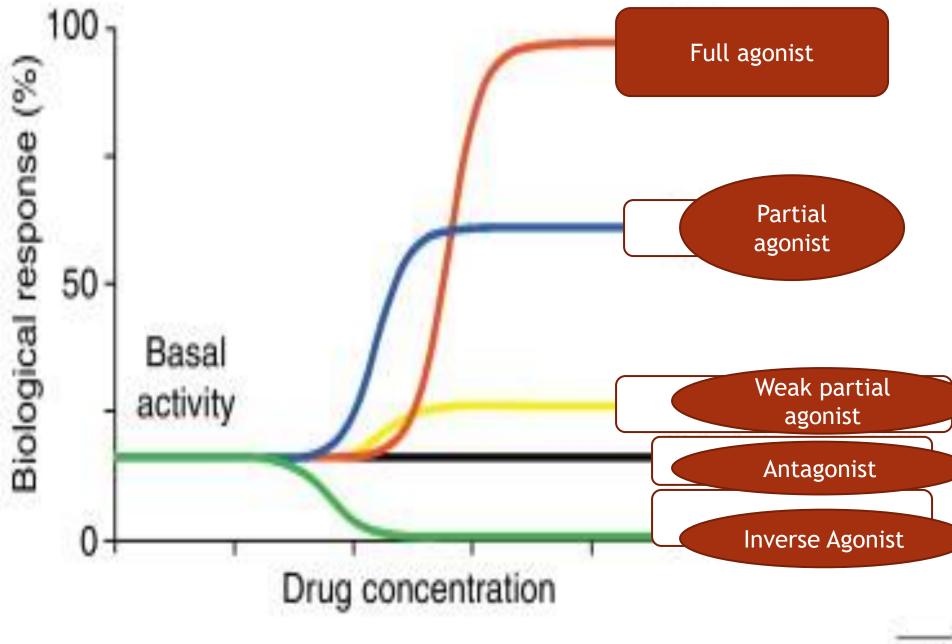
example

Activated charcoal adsorbs toxic substances in case of poisoning.



References

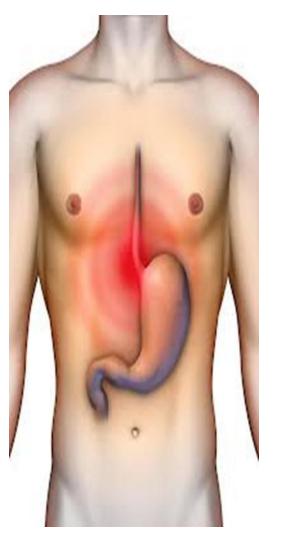
- ▶ 12th edition, by Bertram G. Katzung, Susan B. Masters, and Anthony J. Trevor
- Lippincott Illustrated Reviews: Pharmacology 6th edition (Lippincott Illustrated Reviews Series) Sixth, North American Edition
- Pharmacology Principles and Applications Paperback – April 1, 2016 by <u>Abdul Jalil</u> <u>Popalzai</u> (Author).













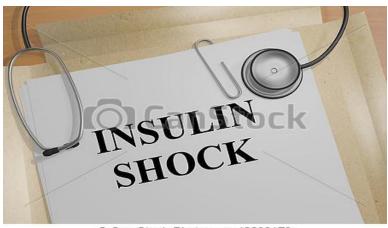
Physphysiologicalantagonism







Chemical antagonism



© Can Stock Photo - csp42209179



Physiological Antagonism

Phenobarbitone increasing the metabolism of warfarin

Warfarin effect being reduced by in its increased metabolism

Pharmacokinetic antagonism

