



METABOLISM OF PROTEINS AND AMINO ACIDS

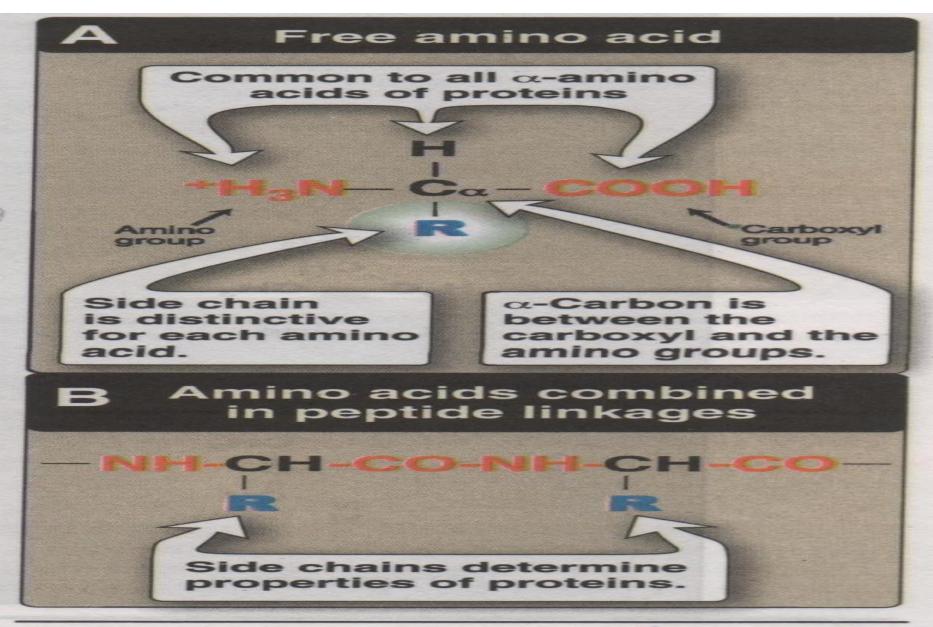
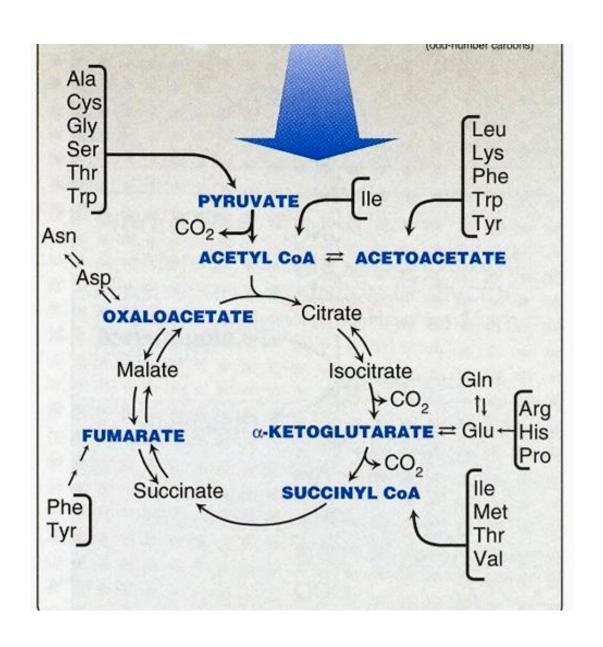


Figure 1.1

Structural features of amino acids (shown in their fully protonated form).



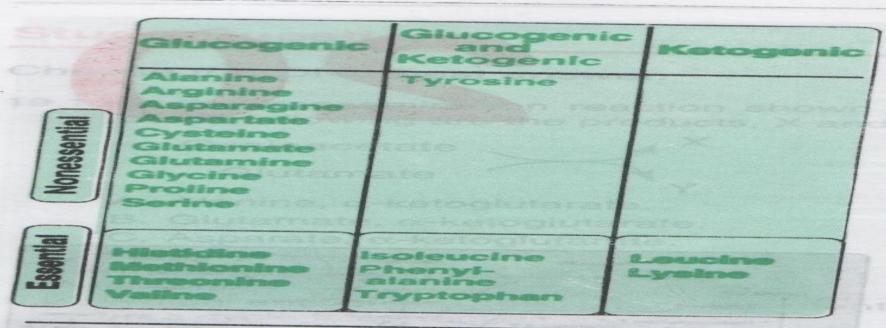


Figure 20.2 Classification of amino acids.

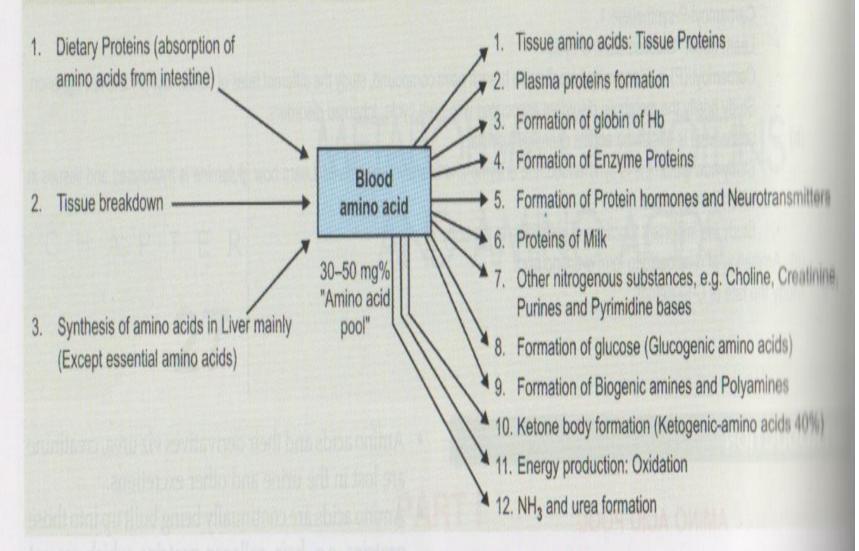


Fig. 27.1: Sources and utilisation of amino acids

CONSEQUENCES OF DIET LOW IN PROTIEN

Deficiency of essential a.acid

Break down of Tissue Protein.

Protein deficiency –KWASHIORKER

CONSEQUENCES OF DIET HIGH IN PROTIEN

Amino group is converted to Ammonia carbon skeleton converted to glucose and Fats.

Protein Degradation

Two major Enzyme systems

- ATP dependent UbiQuitin Proteasome system of Cytosole.
- 2. ATP independent degradative Enzyme system of Lysosome.

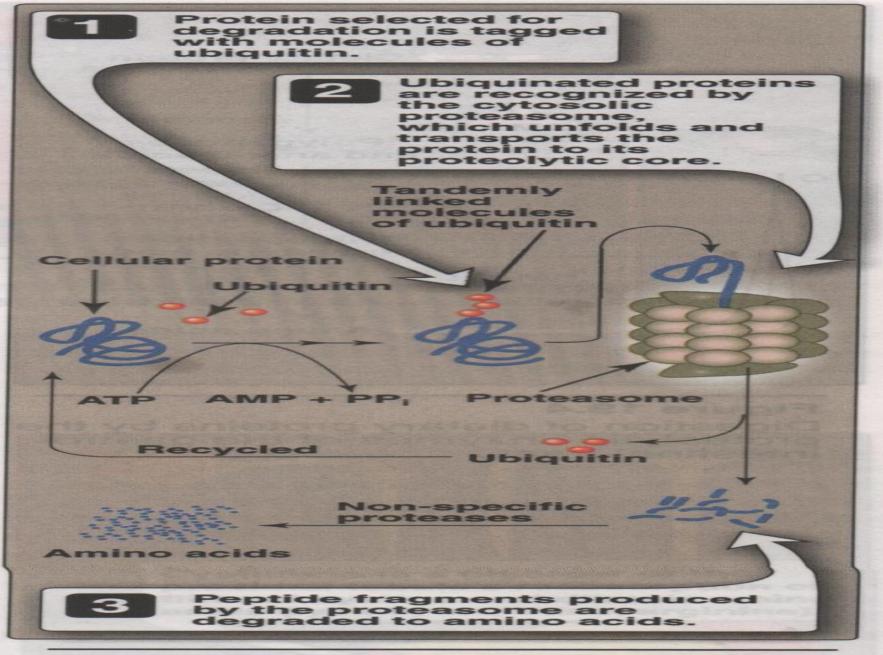


Figure 19.3
The ubiquitin-proteasome degradation pathway of proteins.

Ubiquitin-Proteosome Mechanism

-Ubiquitin –

Small globular non enzymatic Protein

- -Proteosome
- Large barrel-shaped macromolecule, proteolytic complex
- -UbiQuitination: Occure by peptide linkage of carboxyle group of C-Terminal Glycine of UB to amino group of lysine in the protein substrate.
- It is ATP dependant Process.

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Half Life of Proteins

- -Extra cellular protein e.g digestive enzymes and plasma protein = hours or a few days
- Intra cellular proteins e.g collagen = Months or Years
- -Also influence by amino terminal residue e.g serine = More than 20 hours.
- Aspartate = Only 3 Minutes

Biochemical Reactions Of Amino Acids In The Body

Following are various types of reactions involved in the metabolism of amino acids:

- 1. Trans-amination
- 2. Deamination
- 3. Trans-methylation
- 4. Deamidation
- 5. Trans-peptidation
- 6. Decarboxylation &
- 7. Interconversion of amino acids.

1. Trans-amination

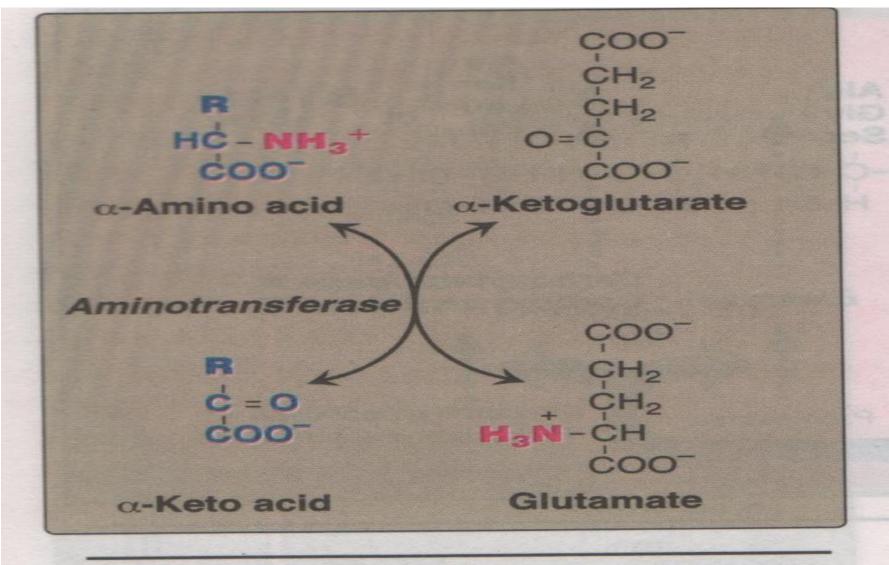


Figure 19.7

Aminotransferase reaction using α-ketoglutarate as the aminogroup acceptor.

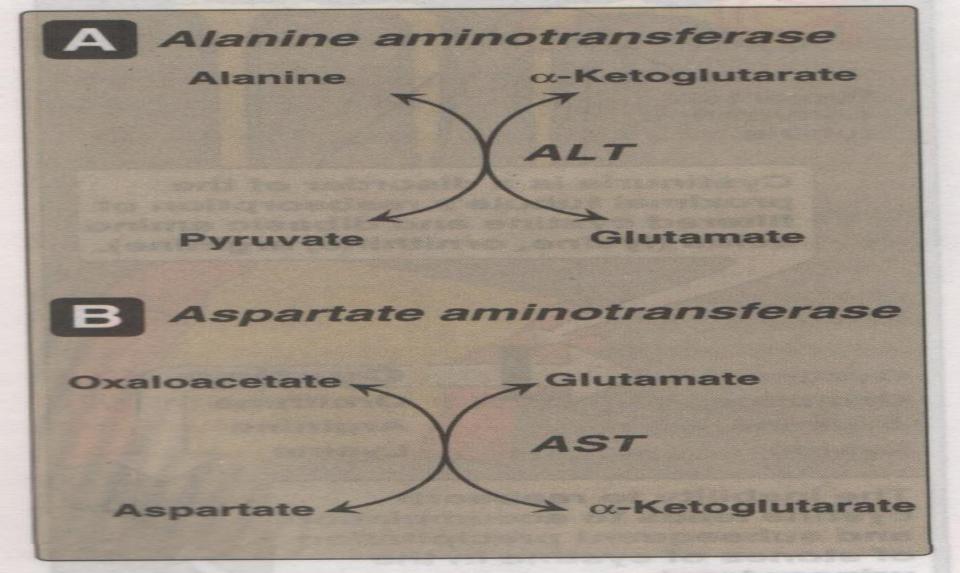


Figure 19.8

Reactions catalyzed during amino acid catabolism. A. Alanine aminotransferase (ALT). B. Aspartate aminotransferase (AST).

Figure 19.9

Cyclic interconversion of pyridoxal phosphate and pyridoxamine phosphate during the aspartate aminotransferase reaction.

[Note: P = phosphate group.]

 All amino acids with exception of lysine and threonine lose their amino group by transamination.

 These 2 amino acids lose their amino groups by deamination.

Diagnostic value of Plasma amino Transferases:-

- Normally Intracellular enzymes
- Elevated level indicates damage to the cell rich in these enzymes.
- AST OR SGOT
- ALT OR SGPT

- (a) Liver diseases. Both elevated
- Also have prognostic value
- (b) Nonhepatic diseases e.g M.I, Muscle disorders.

2. Oxidative Deamination

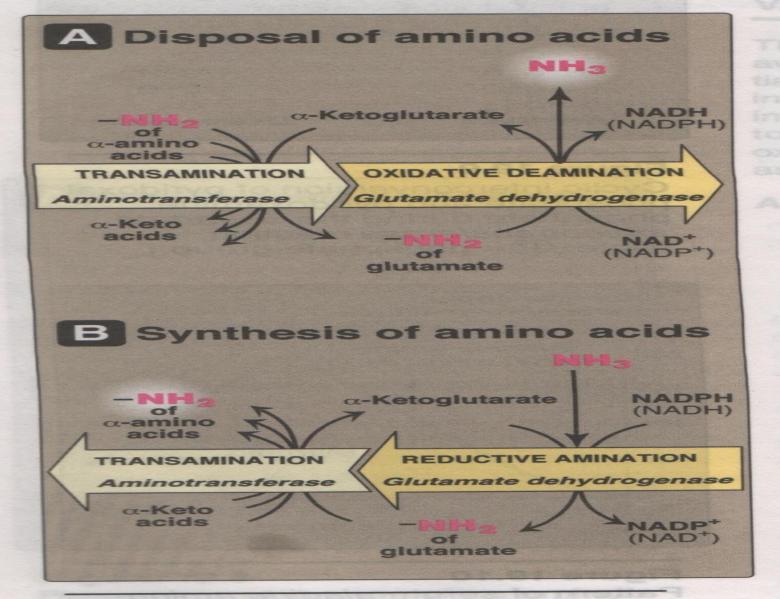


Figure 19.12
Combined actions of aminotransferase and glutamate dehydrogenase reactions.

Transport of ammonia to the liver

- -Two mechanisms
- 1. Combination of ammonia with glutamate to form non-toxic glutamine
- 2. Formation of alanine by transamination of pyruvate.

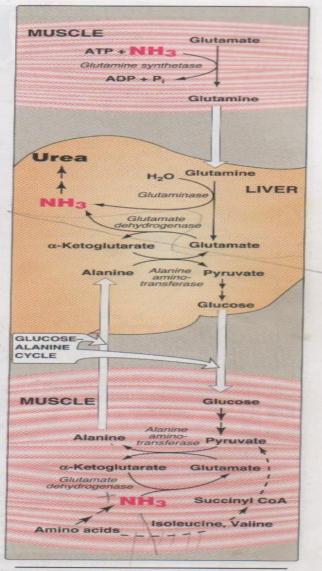


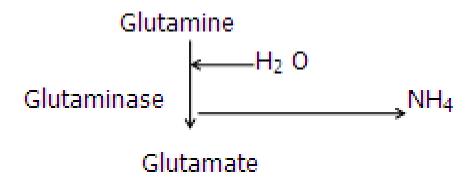
Figure 19.13 Transport of ammonia (NH_3) from muscle to the liver. ADP = adenosine diphosphate; P_i = inorganic phosphate; CoA = coenzyme A.

The transport of alanine
from muscle to liver
results in a reciprocal
transport of glucose to
muscles (Glucose –Alanine Cycle)

MUSCLE Glutamate ATP + NI Glutamine synthetase ADP + P Giutamine Urea Glutamine H-O LIVER Glutaminase Glutamate dehydrogenase α-Ketoglutarate Glutamate Alanine Alanine Pyruvate aminotransferase Glucose GLUCOSE ALANINE ! CYCLE MUSCLE Glucose Atamino amino-Alanine transferase α-Ketoglutarate Glutamate Glutamate denydrogenase Succinyl CoA Isoleucine, Valine Amino acids

SOURCES OF Ammonia

- ${f 1}$. From Amino acid .
- By Transamination and oxidative Deamination
- Quantitatively the most important source
- From Glutamine



- Imp in acid-base balance
- By Bacterial degradation of urea in intestine.
- 4. From Amines by the action of Amine oxidase
- Amines obtained from the diet and Mono amines that serve as Neurotransmitters or hormones
- From Amino groups attached to the purine and purimidine rings.

Metabolic Fate of NH3:-

A Urea Formation:

(B) Formation OF Glutamine:

Some NHB 12 used to Glutamine

aminate Glutamic acid to 7000 Glutamine

E Formation OF Non Essential a.acids:
a Ketoacids are aminated to

gram or amino acids.

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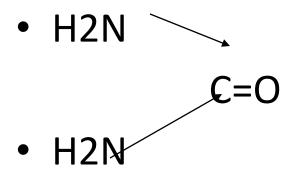
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E Formation OF Non Essential a.acids:
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• Urea Cycle:



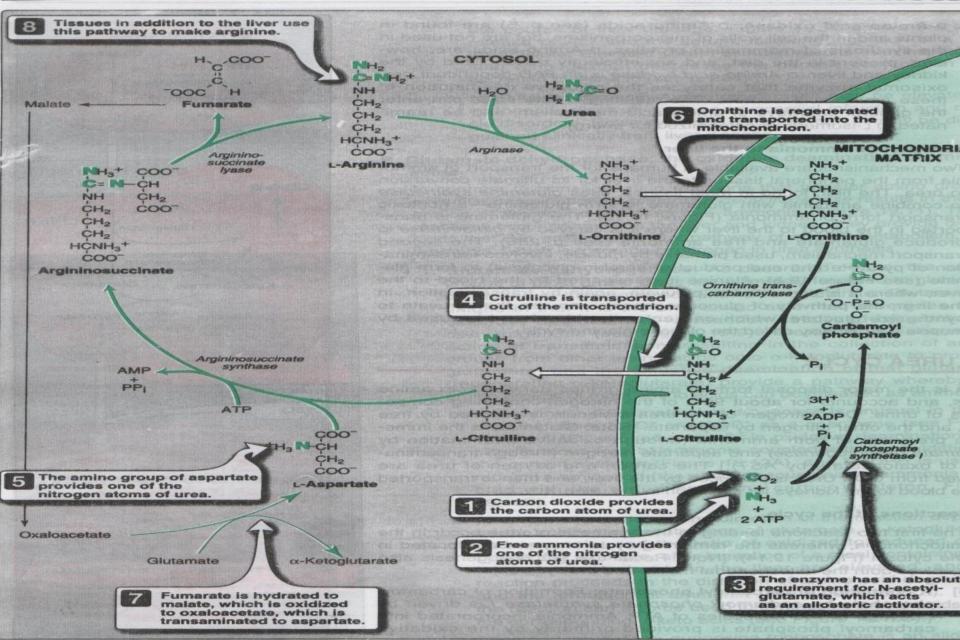
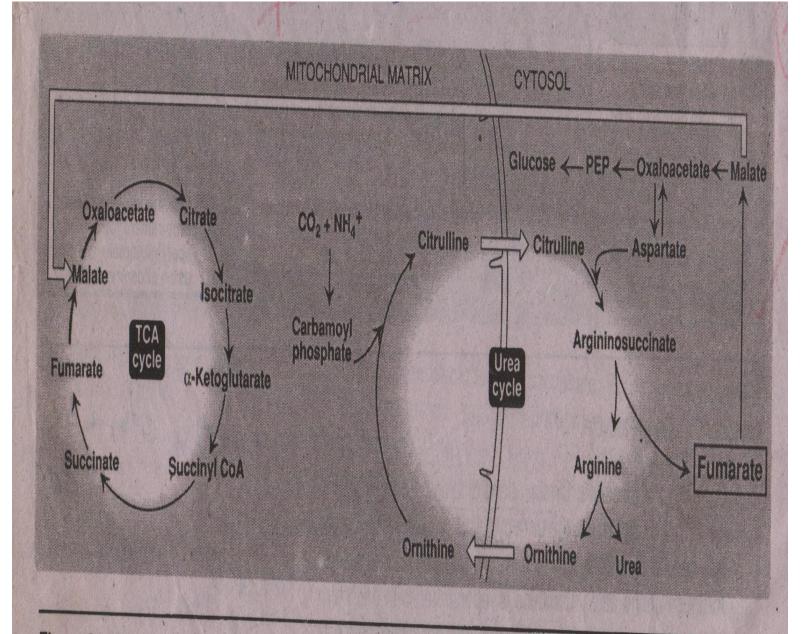


Figure 19.14
Reactions of the urea cycle.

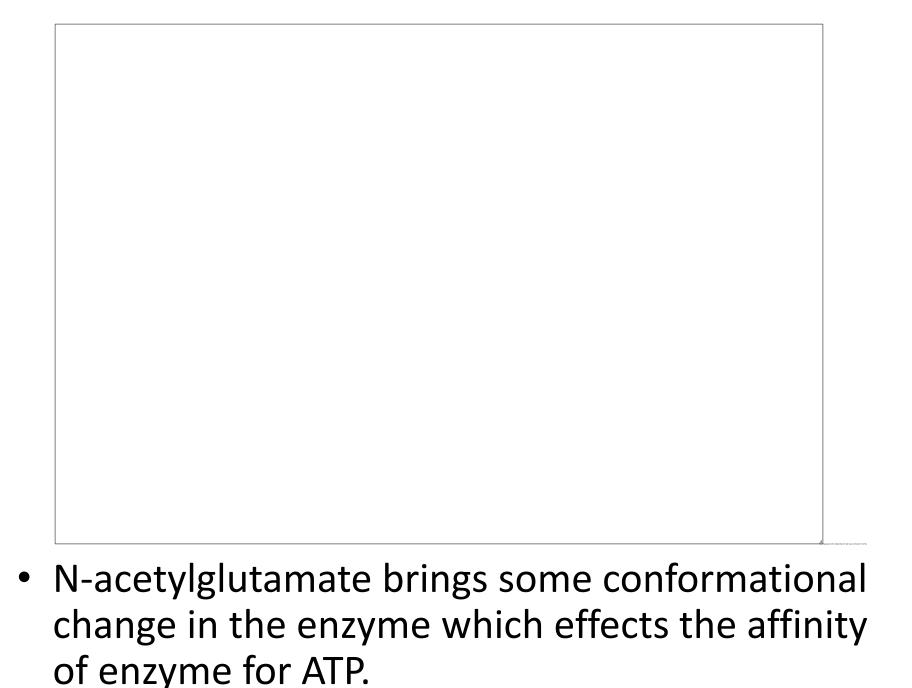


Fate of fumarate produced by the urea cycle.

C. Regulation of the urea cycle

N-Acetylglutamate (NAG) is an essential activator for CPS I, the rate-limiting step in the urea cycle. It increases the affinity of CPS

For ATP



 NAG is synthesized from acetyl CoA and glutamate by Nacetylglutamate synthase In a reaction for which arginine is an activator.

The cycle is also regulated by substrate availability (short-term regulation) and enzyme induction (long term)

Fate of urea

- 1. Mostly filtered and excreted by kidneys
- 2. A small portion, diffuses from blood in to intestine and is cleaved to co₂ and NH₃ by bacterial urease
- ThisNH₃ is partly lost in faces and is partly reabsorbed in to the blood.
- In renal failure, plasma urea level is elevated, so more urea in gut, so more formation of NH₃ causing hyperammonia in these patients.

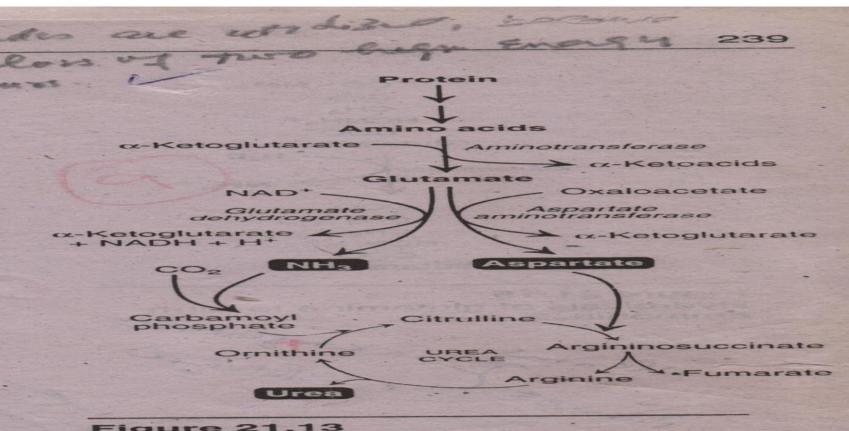


Figure 21.13
Flow of nitrogen from amino acids to urea. Amino groups for urea synthesis are collected in the form of ammonia and aspartate.

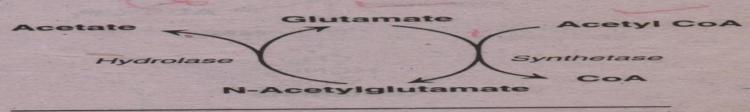


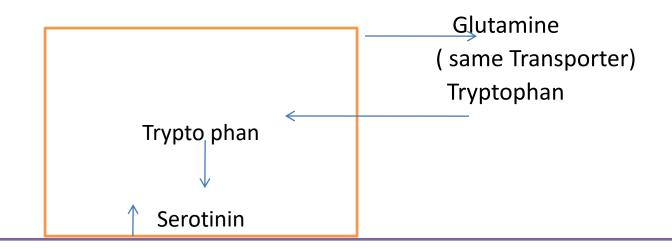
Figure 21.14
Formation and degradation of Nacetylglutamate, an allosteric
activator of carbamoyl phosphate

Significances of urea cycle

- 1. Conversion of Toxic Ammonia in to Non –Toxic urea
- 2. Source of Arginine and Fumarate
- a. Arginine: i. Take part in protein synthesis
- ii. Coverted to NO, which is a powerful vasodilator
- b. Fumarate: i. It is hydrated to malate which is oxidized to oxaloacetate, which in the cytosole is transaminated to asparate and reenter the urea cycle or malate may enter in mitochondria, oxidized to oxaloacetate in TCA cycle, which can be used for gluconeogenesis

Why NH₃ is Toxic

- 1. NH₃ enhances amination of Ketoglutarate of TCA cycle to form Glutamate in brain, so depresses TCA cycle, affecting the cellular respirationespecially the brain, which depends on TCA cycle for its energy.
- 2. Increased NH₃ enhances "Glutamine" formation from Glutamate so decreased Glutamate, so decreased formation of inhibitory Neurotransmitter GABA.
- 3. Increased Glutamine enhances out flow of Glutamine from Brain cell



2. Urea

- Normal value -20-40mg/100ml
- Uraemia Causes
- a) Pre-Renal
- b) Renal
- c) Post Renal

Hyperammonaemia

- 1. Acquired Hyperammonaemia:-
- Cirrhosis of liver
- Renal failure
- 2. Inherited hyperammonaemia:-
- Due to genetic defects in the urea cycle enzymes.
- The symptoms of NH₃ INTOXICATION INCLUDE
- A peculiar flapping Tremor
- Slurring of speech
- Bluring of vision
- Coma and death in severe case

<u>Inherited Disorders Associated with urea cycle</u>

- 1. Hyperammoneamia Type I
- Enzyme deficient : C.P Synthetase I
- Symptoms of Hyperammoneamia
- 2. Hyperammoneamia Type II

INHERITANCE: X-chromosme linked

- Enzyme deficient : Ornithine Transcarbamyolase
- Symptoms of NH₃ Toxicily
- Increased Level of Glutamine, NH₃ and ornithine in blood, urine and CSF.

3. Citrulinemia:-

Inheritance: Autosomal recessive

- Enzyme deficient : Arginin succinate synthetase
- Hyperammonemia
- Increased level of NH₃ and citruline in Blood and CSF
- 4. Arginino Succinic Aciduria:

Inheritance: Autosomal recessive

- Enzyme deficient: Arginino succinase
- Hyperammonemia
- Blood and CSF have increased level of arginino succinate
- 5. Hyperarginimemia: Enzyme deficient: Arginase
- Hyperammonemia
- Increased level of Ariginine in Blood and CSF
- 6. Hyperammonemia may occure due to the deficiency of N-acetyl –glutamate deficiency
- 7. Hyperammonemia may occure due to the deficiency of Ornithine transporter

| Disorder | Deficient Enzyme/ protein | Raised Level |
|------------------------------------|---|---|
| 1. Deficiency of N-Acetylglutamate | N-Acetylglutamate Synthase | Ammonia |
| 2. Hyperammonemia Type – I | Carbamoyl Phosphate Synthetase – I | Ammonia |
| 3. Hyperammonemia Type – II | Ornithine transcarbamoylase | Ammonia + Orn + Uracil + Orotic acid |
| 4. Citrullinemia | Argininosuccinate Synthetase | NH ₃ + Orn + Uracil + Orotic acid + Citrulline |
| 5. Argininosuccinic Aciduria | Argininosuccinase Or Argininosuccinate lyase | NH ₃ + Orn + U + OA + Cit + Argininosuccinate |
| 6. Argininemia | Arginase | NH ₃ + Orn + U + OA + Cit + Argininosuccinate +Arg |
| 7. HHH Syndrome | Ornithine transporter (Responsible for transport of ornithine from cytosol to mitochondria) | Ammonia, Ornithine Homocitrulline (Homocirtullinuria) (When ornithine is not available in mitochondrion, the Carbamoyl Phosphate condenses with lysine to |

- Genetically transmitted (as autosomal recessive genes) (Except type- 2 i.e OT.Carbm deficeincy which is X-linked.)
- Clinical features & treatment of all disorders are similar
- Symptoms and consequences can be minimized when low protein diet is administered as frequent small meals to avoid sudden increase in ammonia level.

Aromatic Amino Acids

1. Phenyl alanine

2. Tyrosine

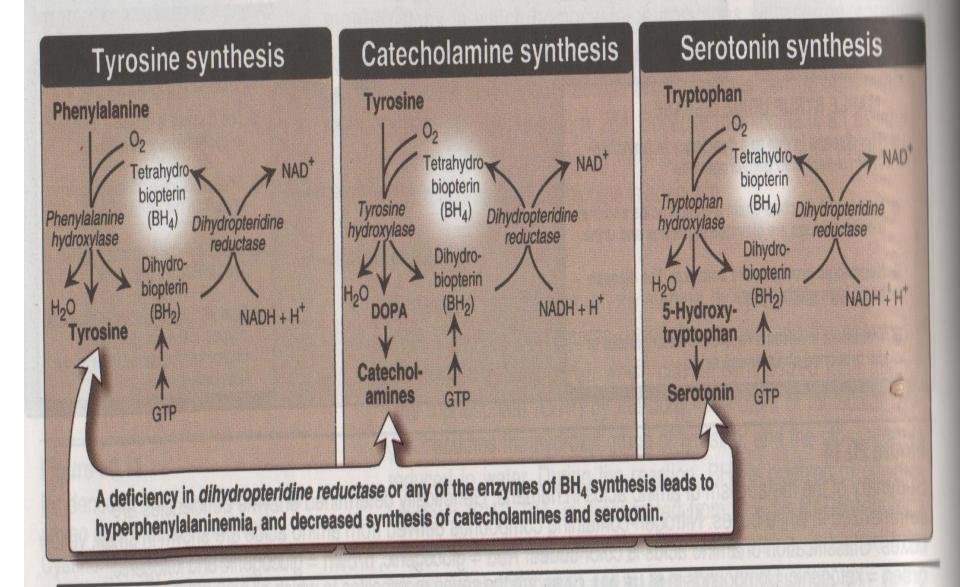


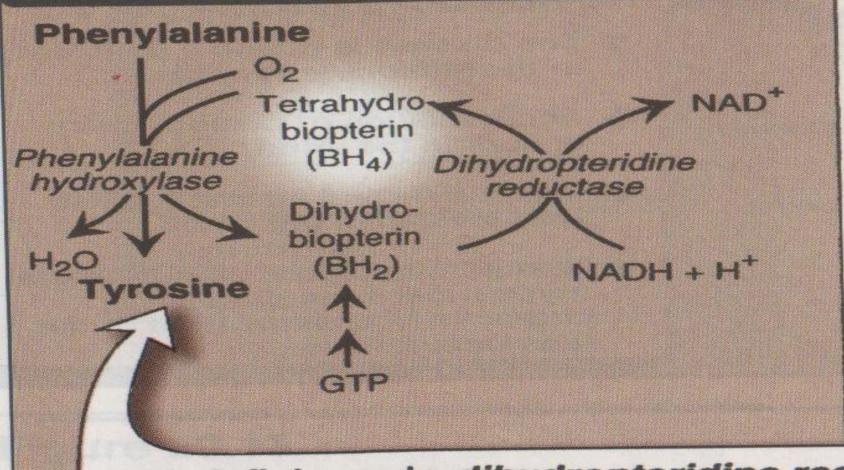
Figure 20.16
Biosynthetic reactions involving amino acids and tetrahydrobiopterin.

Phenyl alanine

- Essential amino acid

- Glucogenic and Ketogenic

Tyrosine synthesis



A deficiency in dihydropteridine rec hyperphenylalaninemia, and decrea

Phenyl ketonuria

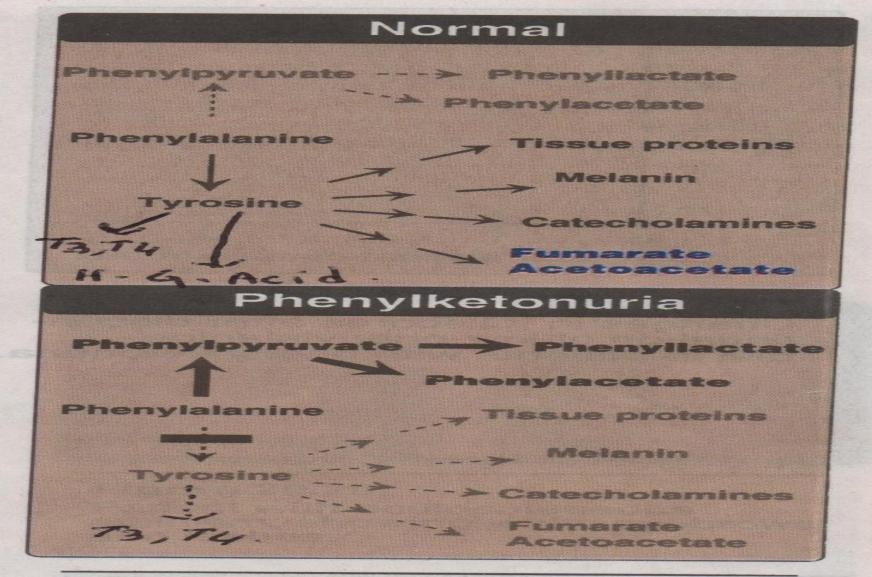


Figure 20.17

Pathways of phenylalanine metabolism in normal individuals and in patients with phenylketonuria.

Phenyl Ketonusia (P.K.U) or Hyper Phonylalaninem ics. Deficiency of P. A. Hydroxylase (TUREI) The Engine Converts P. alanine to Tyrodine which

I's to Precursor of dopamine, Epinep and

I's to Epineph, and thereoid hormones and Melanin - Low Conversion of PhonyPal -> Tyrosine also Occurs = 7 BH4 is not regenerated due to BH4 is also required in biosynth of défective . BH2 Reductage. Neuro Transmillers departine, Novep, Epin, Serotonin Phenylelanine is dwested to is noomed Minor metabolic pathway forming forming Phenyl Provate, Phenylacetate and Phenyl lactate these Metabolites have characteristic MUSTY (MOUSY) Odor.

| Туре | Condition | Probable enzyme defect | Treatment |
|-----------|--|--|------------------------------------|
| I. | Classical type of phenyl ketonuria (PKU) | Phenyl alanine hydroxylase enzyme absent | Low phenyl alanine diet |
| II. | Persistent hyperphenylalaninaemia | Decreased Phenyl alanine hydroxylase enzyme | None but temporary dietary therapy |
| III. | Transient mild hyperphenylalaninaemia | Maturational delay of phenyl alanine hydroxy- lase enzyme | Same as Type II |
| IV. | Dihydropteridine reductase deficiency | Deficient or absent dihydropteridine reductase | Dopa, 5-OH tryptophan, carbi Dopa |
| V. | Abnormal dihydrobiopterin function | Dihydrobiopterin synthesis defect | Same as Type IV |

and usine - C. N. S Symptoms: - TP. Al Impaire brain development

BH4 15 also 169 For Synth of

Neurotransmitters

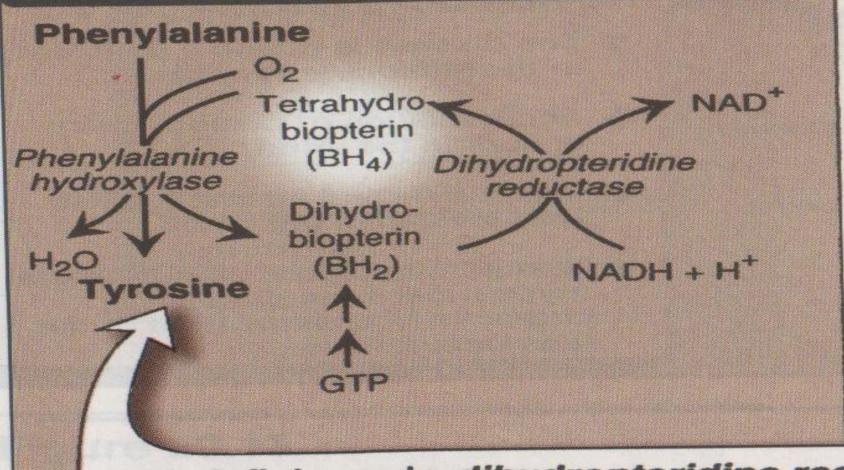
Causing Mental retardation, Seigure, Tremors et.

- Hypo pigmentation 1 -PhonyPelanine is a competitive inhibitor of Tyrosinase in Melanocytes, Causing reduced melanin formation, Causing Fair Skin, Light skin color, Blue Eyes etc. Phenyl acetic acid is conjugated with glutamine and excreted as phenyl acetyl glutamine in urine responsible for " Mousy odour" of urine.

Tyrosine

- Non essential amino acid as it can be formed from Phenyl alanine.
- Degraded to produce end product as Fumarate and acetoacetate
- Glucogenic and Kelogenic.

Tyrosine synthesis



A deficiency in dihydropteridine rec hyperphenylalaninemia, and decrea

Metabolic role of Tyrosine

- Synthesis of Catecholamines.
- Melanin
- Thyroid Hormones
- Tissue Proteins
- Fumarate and Acetoacetate.

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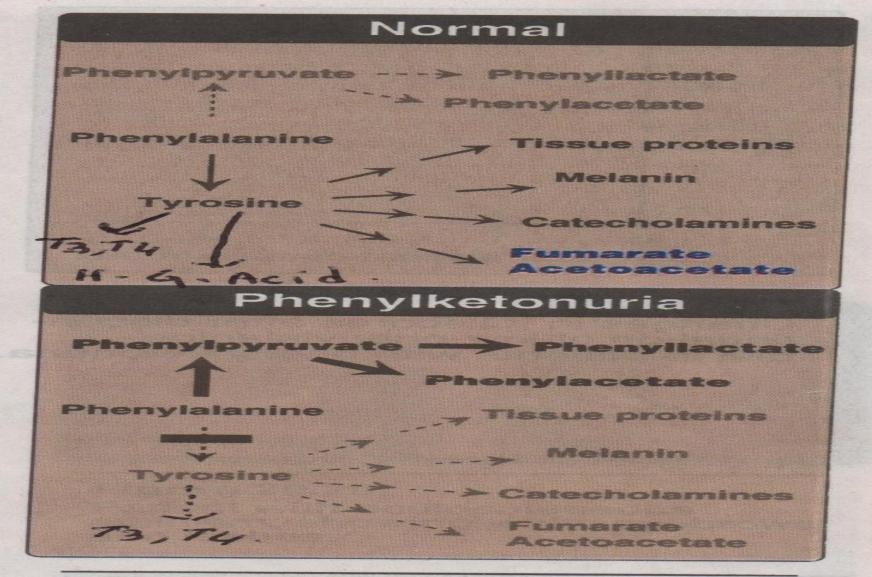


Figure 20.17

Pathways of phenylalanine metabolism in normal individuals and in patients with phenylketonuria.

Tyrosine Catabolism: Tyramine and Phenol in gut 740050,006 TyroSinemia Type II. - There is A Tyrusine plasma lovel Fransami - Eye and Skin Changes - Moderate Mentel letar dation - Turinary Tyrosomi level Pana Hydroxy Phanyl Pyravic acid (cours addition of men offers) Neonatal turosinaemia. vit B12 -> Alkaptonuoia :-Homogentisic acid - Utine has H. G. acid, which when is oxid, is Convetted in brownish black pigmed - so the is darkening of usine then is pigmentation of Connective tussue, Homogentisic due to oxidation of homogentisic acid oxidase COT cleaves benjoquinone acetale - Censonasis aromatic rung Pigment deposition Comser Orthotis. Autosomed recessive disorder. Maleyl acto acetic acid 1 somerization Fumarylacto actic acid I My dolvs/s Tyrosinemia Type I Fumary aceto acetate Hydrolose (Tyrosinosis) Plas Tyr 1: infait exhibit diarrhoen, vomit (Glassenie Rodar) Exerone body) Cabbage like odour - fail to grow death in 6 - 8 months (Hepatic Jail)

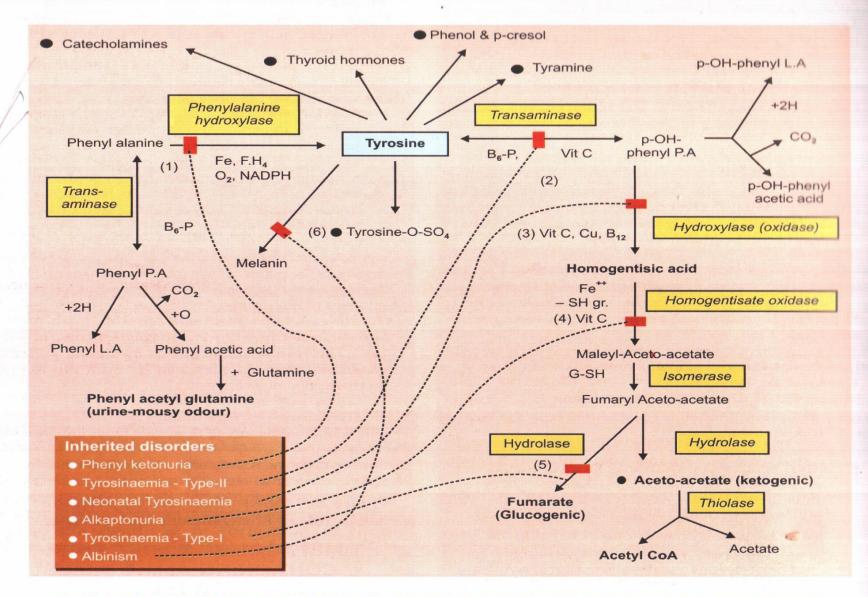


FIG. 27.9: FLOW CHART OF METABOLIC FATE AND METABOLIC ROLE OF PHENYLALANINE AND TYROSINE

Disorders Associated with Phenyl alanine and Tyrosine Metabolism

- 1- Phenyl ketonuria
- 2- Alkaptonuria
- 3- Tyrosinaemia Type-I
- 4- Tyrosinaemia Type-II
- 5-Neonetal Tyrosinaemia
- 6-Albinism

Tyrosine Supplementation Improves

- Mood, Mental ability and sex Drive
- -Also helps in suppressing appetite and reducing body fats
- Good sources are –Any meat or dairy products, eggs, almonds and Bananas.

Catecholamine Synthesis

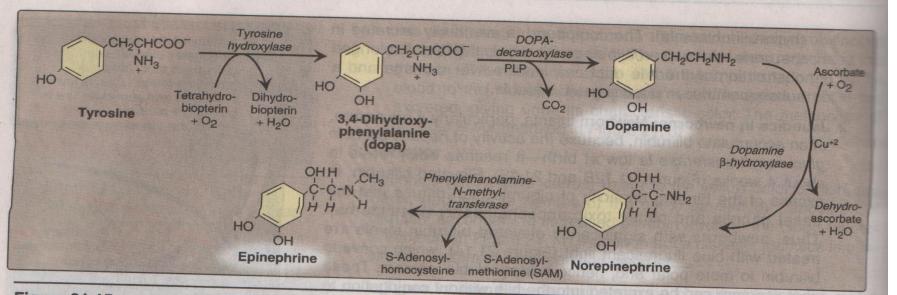
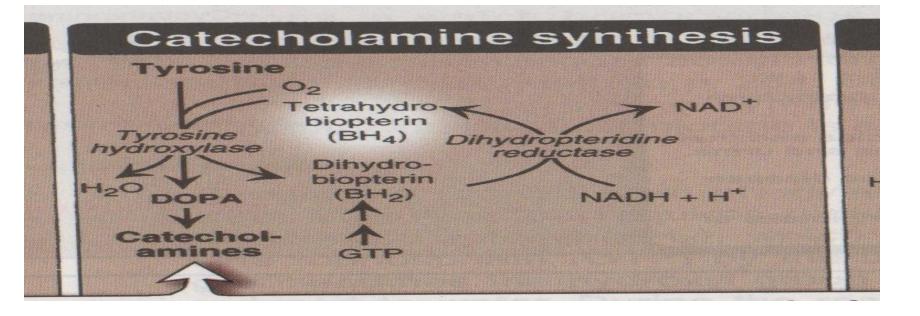


Figure 21.15
Synthesis of catecholamines. PLP = pyridoxal phosphate.



Metabolism of Individual amino acids

Creatine and Creatinine

Creatine

- Normal constituent of the body.

Present in Muscles, Brain, Liver, and blood.

- can occur in free and Phosphorylated form.
- Total amount in body is about 120gm.
- 98 % of total amount is present in muscles of which 80% is in Phosphorylated form.

Creatinine

- Anhydride of creatine. (i.e. removal of one molecule of H2O) (Non-Enzymetic, Irreversible).
- It is in this form that Creatine is excreted.
- Only 2% of Creatine is excreted
 (in the form of Creatinine)
- Formation of Creatinine is Pre requisite for excretion of Creatine.

Normal Creatinine value in blood is 1---- 2 mg/100ml. Used to assess renal functions

In Urine

1.5 ---- 2 gm in 24 hours.

and Creatinine: DG = - 10.3 Kal/mole of creative- P NH₂ ATP= AG= -7-3 K cal from C=NH2 NH CH₂ CH₂ CH2 HCNH3+ HCNH3+ COOT COO Glycine Arginine transferase Ornithine -C=NH2 NH CH2 Kidney . COO-Guanidinoacetate netige group S-Adenosylmethionine Methyltransferase S-Adenosylhomocysteine NHo C=NHo NCH₃ CH₂ COO-Creatine Reversable H20 -Phosphorylation Creatine kinase C=NH ADP + H+ ADP NH NCH3 0-P-0 CH₂ NH2+ C=NH Creatinine Autyaride NCH3 Compound.
(Muscles) CH₂ Creatine COO waste Product -Creatine phosphate Excused in wome **Figure 21.16** Synthesis of creatine. ._ Used To assess NOT not Plasma land - 1-12mg de Rand Functions a in Kidney Diseases

Formation of creatinine

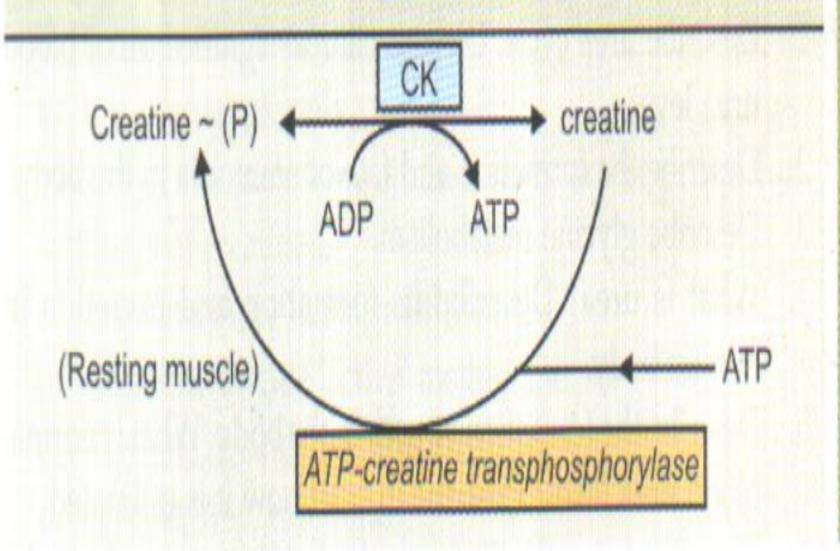
Characteristics of the above reaction

- Reaction is irreversible
- It is non-enzymatic
- Creatinine has ring structure.

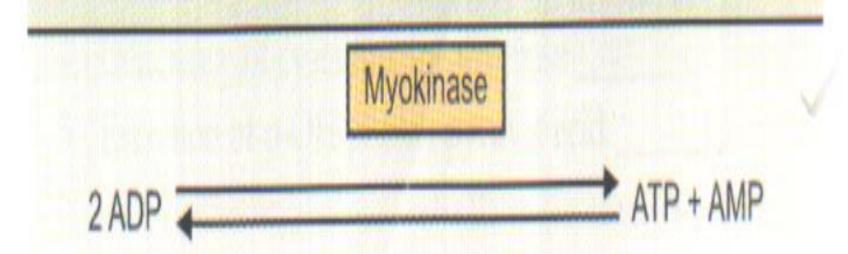
Occurrence and Distribution:

A. Creatine: It is a normal constituent of the body. It is present in muscle, brain, liver, testes and in blood. Can occur in free form and also as phosphorylated form. The phosphorylated form is called as creatine-PO₄ or phosphocreatine or Phosphagen. Total amount in adult human body is approximately 120 gm. 98 per cent of total amount is present in muscles, of which 80 per cent occurs in phosphorylated form, 1.3 per cent in nervous system (brain) and 0.5 to 0.7 per cent in tissues.

Löhmann Reaction



Myokinase Reaction



6 Decarboxylation Reactions

- Reactions by which Co₂ is removed from CooH group of an a. acid by enzyme, Decarboxylase, present in liver, kidneys, brain and also in micro organism of intestinal tract.
- The enzyme require pyridoxal phosphate as coenzyme.
- The enzymes removes Co₂ from CooH and convert the a.acid is corresponding amines.
- (Amino acid)

Examples are :-

- (i) Histidine _____ Histamine + CO₂
- (ii) 5-Hydroxytryptophan 5-Hydroxytryptamine + CO₂
- (iii) Lysine \longrightarrow Cadaverine + CO₂
- (iv) Ornithine \longrightarrow Putrescine + CO₂
- (v) Tyrosine \longrightarrow Tyramine + CO_2
- (VI) Dihydroxy-phenylalanine (DOPA)——→ Dopamine + CO₂
- (VII) Glutamic Acid → GABA + CO₂

| | | Table 27.1: Biogenic amines and their | functions |
|----------------|----------------------------------|---|---|
| Nos | Amino acids | Amine | Biologic importance - |
| 1. | Tyrosine | · Tyramine | Increases blood pressure (Vasoconstriction) Contracts uterus |
| 2. | Tryptophan | · Tryptamine | Tissue hormone: A derivative 5-OH |
| | | | Tryptamine (Serotonin) Vasoconstriction |
| 3. | Histidine | 5-methoxy Tryptamine (Melatonin) Histamine | • BP↑ |
| 4. | Serine | Ethanolamine | Pepsin ↑ Forms choline by three methylations |
| 5. 6. 7. | Threonine Cysteine Aspartic acid | Propanol amine β-mercaptoethanolamine β-alanine | Constituent of Phospholipid like cephalin Constituent of Vit B₁₂ Constituent of coenzyme A Constituent of pantothenic acid (coenz. A) |
| 8. | Glutamic acid | · γ-amino butyric acid | As a constituent of dipeptide carnosine and Anserir Presynaptic inhibitor in brain. |

Forms a bypass in TCA cycle (GABA-shunt)

· Constituent of Bile acid taurocholic acid

· Product of Putrefaction in the gut

· Product of Putrefaction in the gut

Product of Putrefaction in the gut

· Precursor of Epinephrine and Nor-epinephrine

(GABA)

Dopamine

· Putrescine

Agmatine

Cadaverine

Taurine

3,4,-di-OH-phenylalanine (DOPA)

Cysteic acid

Lysine

Ornithine

Arginine

10.

11.

· 12.

5 13.

Histaduie > Histamine

- 1 Formation as a result of study of bees and Allergic conditions.

- 9+ Causes vasodilitation,
redness, and swelling.

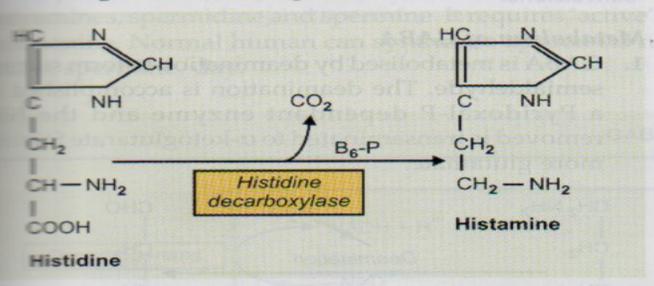
- 1 Histamine 15 also formed in allergic Reactions.

Histamine

Histamine is formed by decarboxylation of amino acid Histidine" by the enzyme *Histidine decarboxylase* aromatic L-amino acid decarboxylase in presence B₂-PO₄.

Formation

Mast cells are the chief source of histamine in the issues and histamine constitutes about 10 per cent of the weight of mast cell granules.



- Also produced by gastric mucosa cells and histamimergic neurones of the central nervous system.
- Basophils are the chief source of histamine in the circulating cells.

Local Action of Histamine

Upon SC injection of hsistamine, it causes (i) pruritus, • erythema, (ii) circumferential flare and a central raised (iii) wheal (wheal and flare).

Blockers of Histamine (Antihistaminics)

 Blockers of H₁ receptors: The anaphylactic reaction can be minimised by pharmacological agents, e.g. Promethazine and Mepyramine which block H₁ receptors.

 Blockers of H₂ receptors: 'Cimetidine' is used to reduce the gastric acidity in peptic ulcer patients, it is blocker of H₂ receptor.

Actions through H₁ receptors

Actions through H₂ receptors

- Contracts smooth muscle
 Produces bronchodilation including airways and the GI tract

- Increases venular permeability
- Increases vasopermeability and dilation
- Induces nasal mucus production
- Induces airway mucus production
- Causes pruritus, with cutaneous vasodilation
- Also causes pruritus with H₂ receptor, stimulates gastric acid secretion. HCl ↑ and pepsin ↑

APPLIED CLINICAL ASPECT

Elevated plasma levels of histamine have been demonstrated in:

- Patients with anaphylaxis, provoked by exercise or antigen.
 Such reactions are related to the explosive liberation of histamine caused by entrance of the sensitizing substances in the tissues.
- During spontaneous episodes of increased symptoms in patients with "mastocytosis", mast cells tumor.
- During experimentally induced angio-oedema in patients with cold urticaria.
- In patients with antigen-induced bronchial asthma.
- Also formed in injured tissues. Excessive liberation of histamine may be related to traumatic shock.
- Histamine markedly depresses blood pressure ↓ and large doses may cause extreme vascular collapse.
- After challenge by specific antigens in patients with 'atopy'
 histamine demonstrated in nasal lavage fluid and skin
 blister fluid.

Local Action of Histamine: Upon SC injection of histamine, it causes • pruritus, • erythema, • circumferential flare and a central raised • wheal ("wheal and flare").

Glutathions

Gluta Thione

- Tri-Reptide OF 3 aminoacids.

= Glutamie acid

= Cysteine

Glycine.

Synthesis :.

Glutamic Acid + - Cysleine ADP +P: 4 Synthase 7-L- Glutamy - L- cysteine ATP (Glutathione Synthetase) -- glatamy - custeing -(P) Calutalhione + Pi

Suportant reducing agent in The Tissues.

NADPHIN GlutaThione

oxidized Glutathione is harmful to the Tissues expecially R.B.C and lone Protein, and is Converted to reduced Glutalthione, which is required for integrity of R.B.e memb = leve foolen

Catabolism a degradation alutathione - Insulin Transhydrogenase

= Many - Sti group containg Endymes e. q. glycereldehydo-3-P-d.H are Protected by Glutathione against oxidation.

114

- G-SH is required as co-onzyme 3 in reaction Melhionine > 5-adenasy Melhionine - GI-SH is required as coenzyme in cyclo-oxygenese system required - For P. G. - Synthesis. = Toxic Radicals or Reactive Intermediates
Formed from Molecular oxygen are Super oxide Hydrogen Hydroxyl Peroxide Glut-Perox adical - Various Intracellular reducing agents (Antioxidante) are Vitamin - E B - carotene. = Ascorbate alutalhione HADAN = = Superoxide dismutase = Catalase = Glutathione Peroxidose.

Serine

= Non Essential, hydroxy a. acid.

Tissue Proloin. custoine Custine sphingol Segine

Sphingomylein

Eth

Clusine Ethanolamine (three)

Ethanolamine (tholin. > cephalin (Phosphatidyl-Ethan alamine) Glucogenice [P. A. · Hydroxy group of serine in an enzyme Protein is phosphosylated/dephosphosylated to from active/inactive forms of The engyme.

NITRIC OXIDE

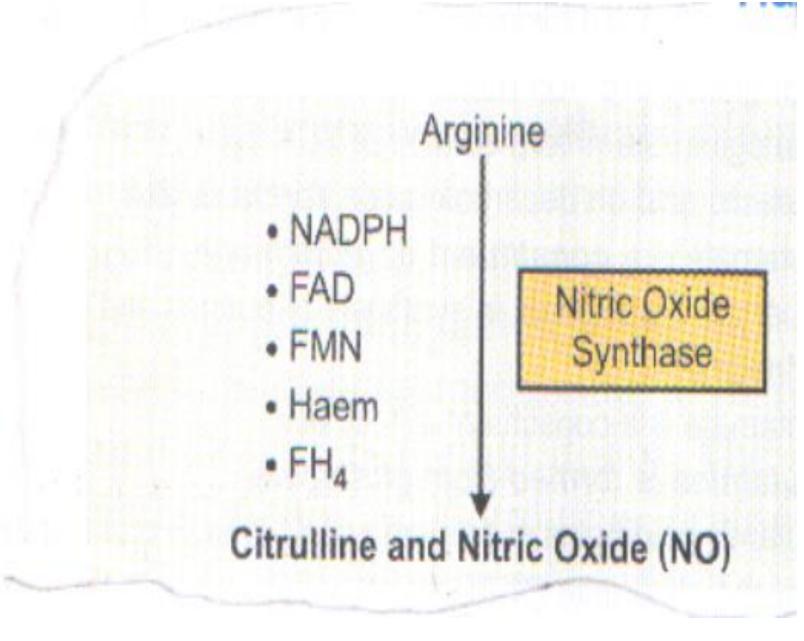
ROLE OF NITRIC OXIDE

Nitric oxide (NO) is formed in the body from amino acid arginine

It is a wonder molecule having diverse biological functions like PGs. Endothelium derived relaxing factor (EDRF) which produces vasodilatation is now proved to be nitric oxide.

Formation of NO

Arginine is acted upon by an enzyme called nitrogen oxide synthase, a cytosolic enzyme and converts arginine to citrulline and nitric oxide (NO).



Duration of action of NO: Nitric oxide formed in the tissues has a very short half-life, approximately 3 to 4 seconds because it reacts with oxygen and superoxide. The product of the reaction with superoxide is Peroxynitrite (ONOO-), which decomposes to form the highly reactive OH radical.

Functions of Nitric Oxide

- It acts as a vasodilator and causes relaxation of smooth muscles.
- It has important role in the regulation of blood flow and maintaining blood pressure.
- It is involved in penile erection.
- Acts as a neurotransmitter in the brain and peripheral autonomic nervous system.
- May have also role in relaxation of skeletal muscles.
- Inhibits adhesion, activation and aggregation of platelets.
- May constitute part of a primitive immune system and may mediate bactericidal actions of macrophages.
- Low level of nitric oxide may be involved in causation of pylorospasm of infantile hypertrophic pyloric stenosis.

CLINICAL ASPECT

- Nitroglycerine: The important coronary artery vasodilator used in Angina Pectoris acts to increase intracellular release of EDRF (now proved to be NO) and cGMP[↑].
- In septic shock: Bacterial lipopolysaccharide present in blood causes uncontrolled production of NO leading to dilatation of blood vessels and lowering of BP.
- In eclampsia and pre-eclampsia: The hypertension is due to decreased production of nitric oxide (NO) due to probably formation of ADMA (asymmetric dimethyl arginine).
- Iron supplements: Iron supplements can dramatically reduce dry cough symptoms in heart patients. Cardiac patients using ACE inhibitors, widely prescribed for hypertension, heart failure and other cardiac conditions

often suffer from a dry cough. It is the biggest reason for people stopping taking their medication. Iron supplements act by decreasing the production of Nitric oxide, which is linked to inflammation of the bronchial cells in the lungs.

GLYCINE METABOLISM

(Non-Essential, glucogenic)

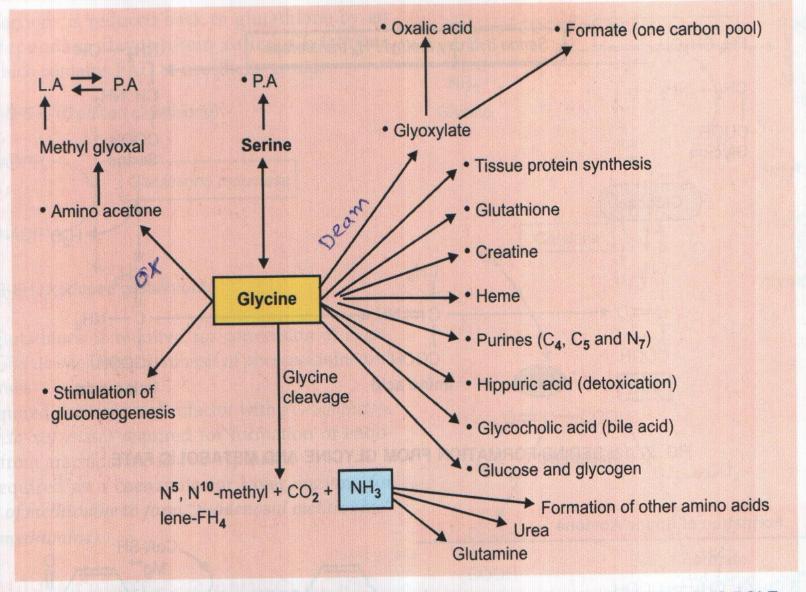


FIG. 27.13: FLOW CHART OF GLYCINE, SHOWING METABOLIC FATE AND METABOLIC ROLE

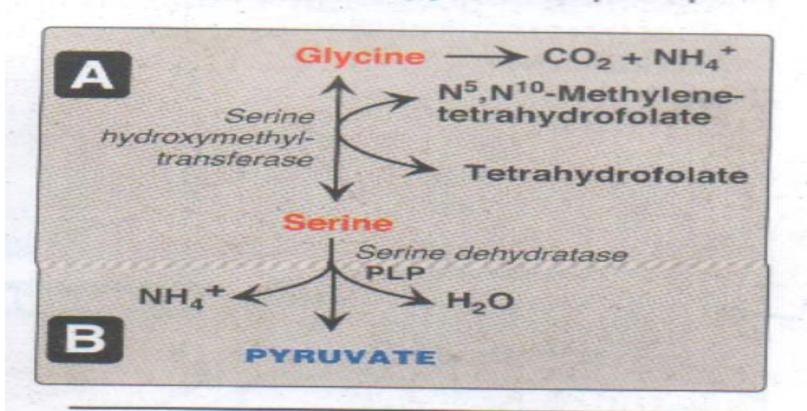


Figure 20.6

A. Interconversion of serine and glycine and oxidation of glycine.

B. Dehydration of serine to form pyruvate. PLP = pyridoxal phosphate.

| Glutamine |
|--|
| NH3 Glutamic |
| Gluidmira Acid. |
| Glutamine Synthetase Glutamine Synthetase |
| |
| - Detoxification Process. |
| - Delovide - 6-12 mg/d1. |
| |
| Biochem Functions: |
| Biochem Functions:. @ Fouclase - 6-P + Glutomine - Glucosamine - 6-Po + Glutomate. |
| |
| 3 Formation of M- Formyl glycinamide (Rumi Muchestide) Cornation of M- Formyl glycinamide (Rusinialmic Muchest) |
| 3 Ecrmation of N- Ecamil Blacinamige (Companie) |
| @ Formatte |
| (5) Glutamine Saves brain From Toxic |
| extects of NH3. |
| |

Functions of Glutamate

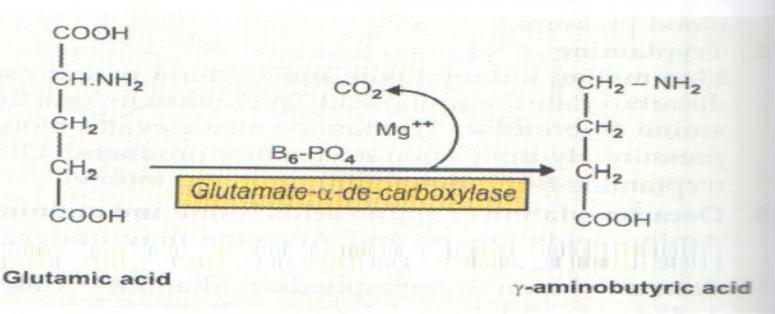
- Participates in Transamiantion reactions in nitrogen disposal
- Component of Glutathione and FH₄
- Most abundant excitatory neurotransmitter in the brain
- Decarboxylated to GABA in neurons

GABA

5. γ-aminobutyric Acid (GABA) formation:

Decarboxylation of glutamic acid produces γ-aminobutyric acid (GABA).

- Reaction is irreversible
- glutamate α-decarboxylase is the enzyme which catalyses the reaction.
- It requires B₆-PO₄ as coenzyme and Mg⁺⁺ as cofactor.



Site of formation

Site of formation

- Principally formed in CN system in the gray matter
- Kidneys

Function of GABA

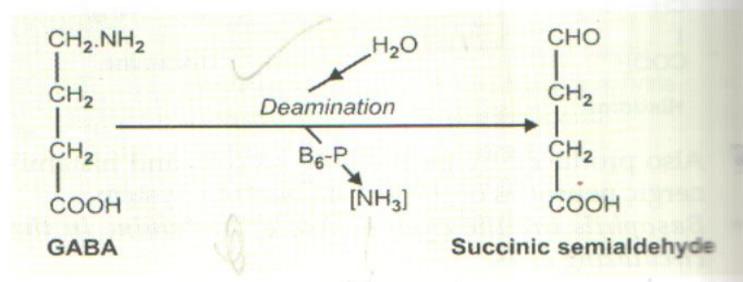
- GABA is known to serve as a normal regulator of neuronal activity being active as an inhibitor (pre-synaptic inhibition).
- It is released at the axonterminals of neurons in grey matter and acts as inhibitory neurotransmitter by enhancing K⁺ permeability of postsynaptic membranes.

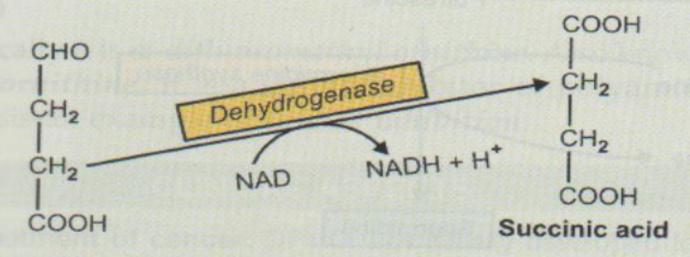
CLINICAL ASPECT

Vit B₆ deficiency in children may be responsible for some of the cases of infantile convulsions. B₆-deficiency causes less formation of GABA leading to neuronal hyperexcitability and convulsions.

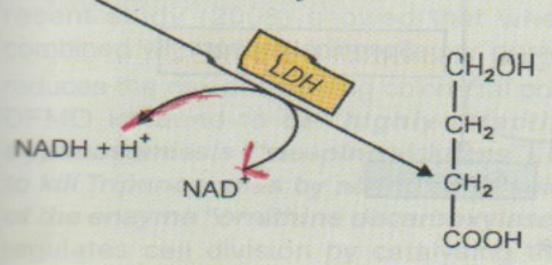
Metabolism of GABA

GABA is metabolised by deamination to form succinition semialdehyde. The deamination is accomplished by a Pyridoxal-P dependant enzyme and the NH removed is transaminated to α-ketoglutarate forming more glutamate.



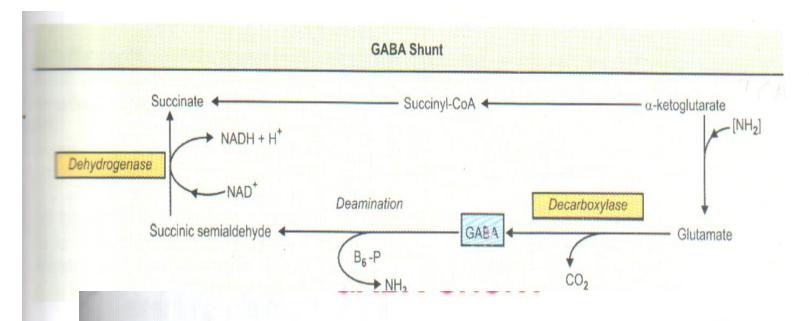


Succinic semialdehyde



y-OH-butyrate

GABA SHUNT

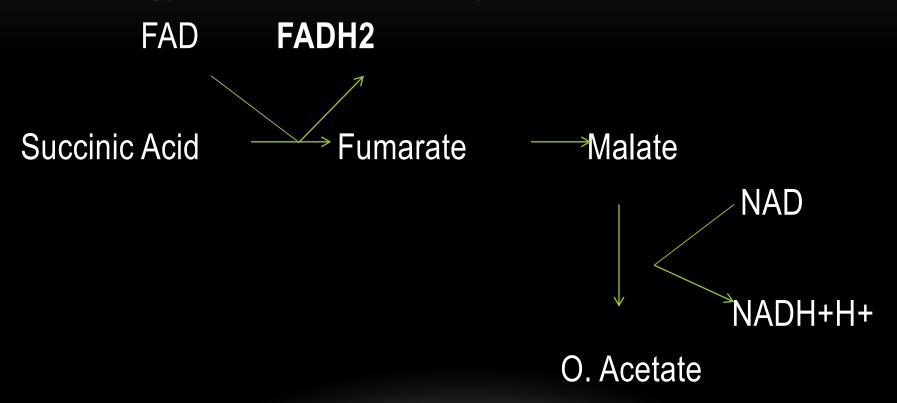


GABA by its conversion to succinic acid can form a "bypass" in TCA cycle and this is called as GABA-shunt

- GABA shunt is a closed loop process, with the dual purpose of producing and conserving the supply of GABA.
- GABA is present in high concentration in brain.

GABA SHUNT

GABA on breakdown is converted in to succinic acid, so it is turned in to energy as it enters in T.C.A cycle.



FADH2 and NADH+H+ will give energy in E.T.C. so by shunting GABA in T.C.A, there is not totally loss of GABA but it gives energy.

Brach Chain Amino Acids

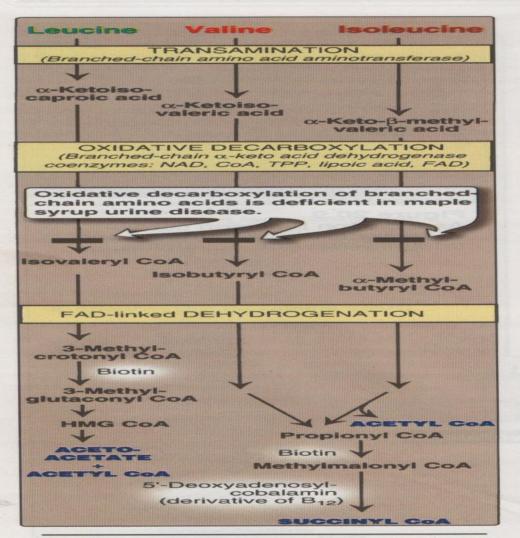


Figure 20.10

Degradation of leucine, valine, and isoleucine. TPP = thiamine pyrophosphate. [Note: 3-Methylcrotonyl CoA carboxylase is one of four biotin-requiring carboxylases we have encountered. The other three are pyruvate carboxylase, acetyl CoA carboxylase, and propionyl CoA carboxylase.]

Maple Syrup Urine Disease (Burnt Sugar Smell)

Maple Syrup Urine Disease

An inherited disorder of branched chain amino acids.

Enzyme defect: Absence of α-ketoacid decarboxylase a greatly reduced activity of the enzyme. As a result the conversion of all three branched chain α-ketoacids to CO₂ and acyl CoA-thioesters is interferred with.

Clinical features: The disease is evident by the end of first week of extrauterine life. Infant does not take feed and man vormit, poor muscle tone. The patient may exhibit lethargy and

surviving children and mental retardation. Without treatment, death usually occurs by the end of the first year of life.

Blood: Plasma levels of the branched chain amino acids eucine, isoleucine, valine and their corresponding α-ketoacids greatly elevated.

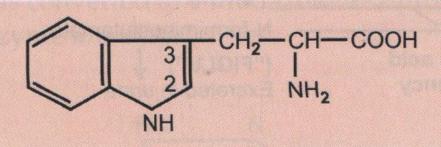
Their corresponding α -ketoacids are excreted. Hence it is called as **Branched-chain ketonuria**. Small amounts of manched-chain α -OH-acids, formed by reduction of α -keto-acids are also excreted in urine. The urine has characteristic adour, which resembles that of **maple syrup or burnt sugar**, tence the name.

TRYPTOPHAN

Points to remember

- It is an essential amino acid. Omission of tryptophan in diet of man and animals is followed by tissue wasting and negative nitrogen balance.
- It is both glucogenic and ketogenic.
- Tryptophan can synthesize niacin (nicotinic acid), a vitamin of B-complex group.
- It is a hetero cyclic amino acid and chemically it is " α -amino- β -3-indole propionic acid". It is the only amino acid with an indole ring.

Structure is shown below:



Tryptophan

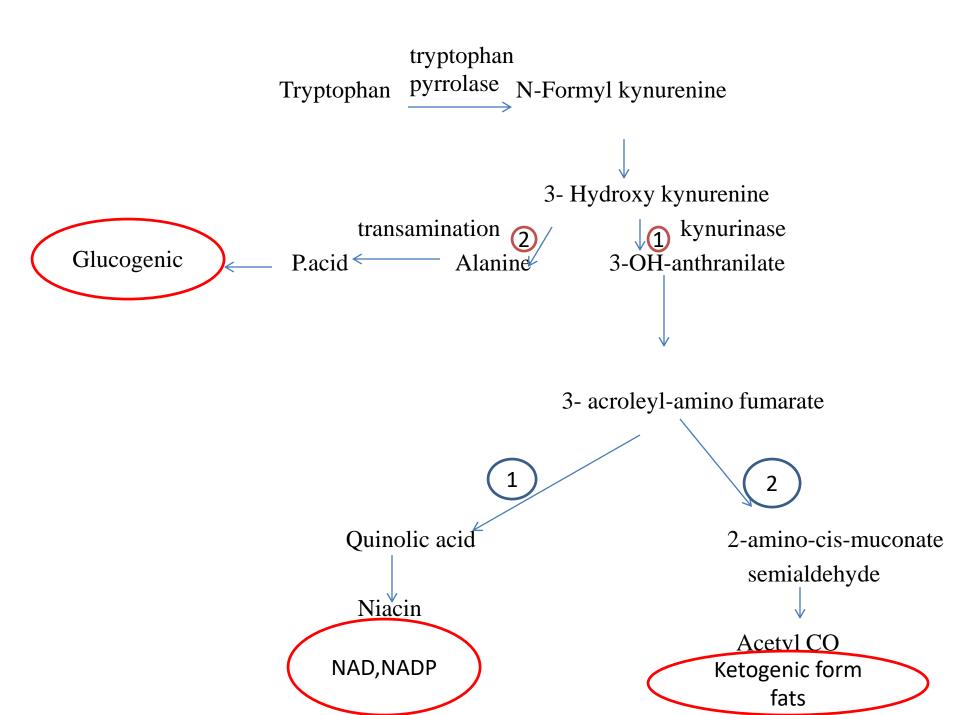
Precursor of

- Serotonin
- Melatonin
- On catabolism by "Kynurenine anthranilate" Path

Tryptophan metabolisim

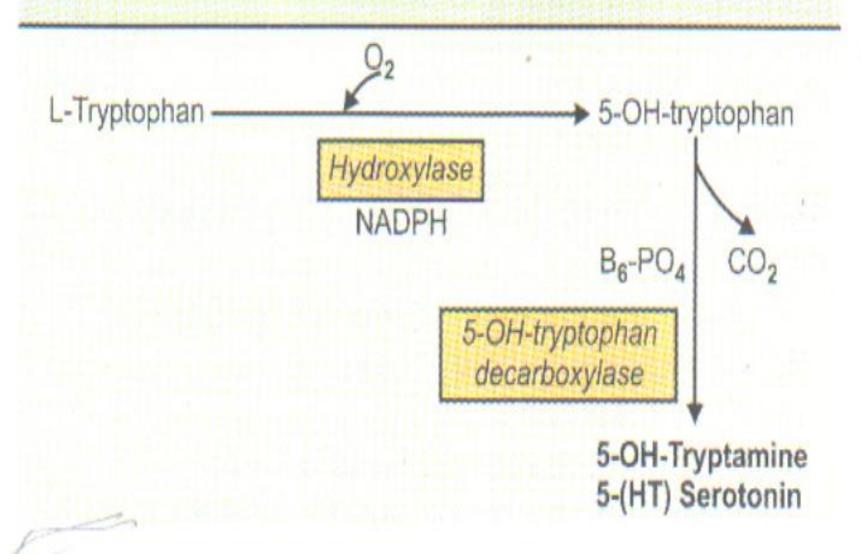
Tryptophan by kynurenine pathway forms

- Glucose
- Fats
- Niacin (Nicotinic Acid)
- NAD
- NADP





Formation of Serotonin



Serotenin:

Functions:

- 1. Vaso contrictor
- 2. Smooth Muscle Contraction
- 3.In Brain acts as a Neurotransmitter

Its excess in the Brain stimulates cerebral activities

Its deficiency produces Depression.

After release, serotonin is destroyed by an enzyme MAO

(Mono Amine Oxidase)

MAO Inhibitors are used to treat the patients of depression

These drugs prevent the destruction of serotonin by MAO, So serotonin is accumulated in the brain.

CLINICAL ASPECT

INHERITED DISORDER

Hartnup Disease

A hereditary disorder associated with defective tryptophan metabolism. Named after the family in which it was discovered.

Biochemical defect: It is not known exactly. Probably impaired formation of "transport proteins" for tryptophan and neutral amino acids in intestinal mucosal, renal tubular epithelial cells and the brain. There is defective intestinal and renal transport of tryptophan and other neutral amino acids.

Clinical features: These are characterised by:

- Mental retardation.
- Intermittent cerebellar ataxia and other neurological symptoms.
- Pellagra-like skin rash—cutaneous hypersensitivity to sunlight.

Blood: Plasma level of tryptophan and other neutral amino acids are reduced.

Faeces and Urine: The neutral amino acids, including tryptophan are excreted in urine and faeces, at least 5 to 10 times of normal average. Faecal excretion of tryptophan is specially marked after a "loading" dose of tryptophan given orally.

Urine: Also shows greatly increased amounts of Indoleacetic acid.

Note

There is decreased synthesis of serotonin and nicotinic acid, which accounts for neurological symptoms and pellagra like rash respectively.

Sulphur containing amino acids

- 1. Methionine
- 2. Cystein
- 3. Cystine

METHIONINE METABOLISM

(Sulphur containing ,essential, glucogenic)

METABOLIC ROLE OF METHIONINE

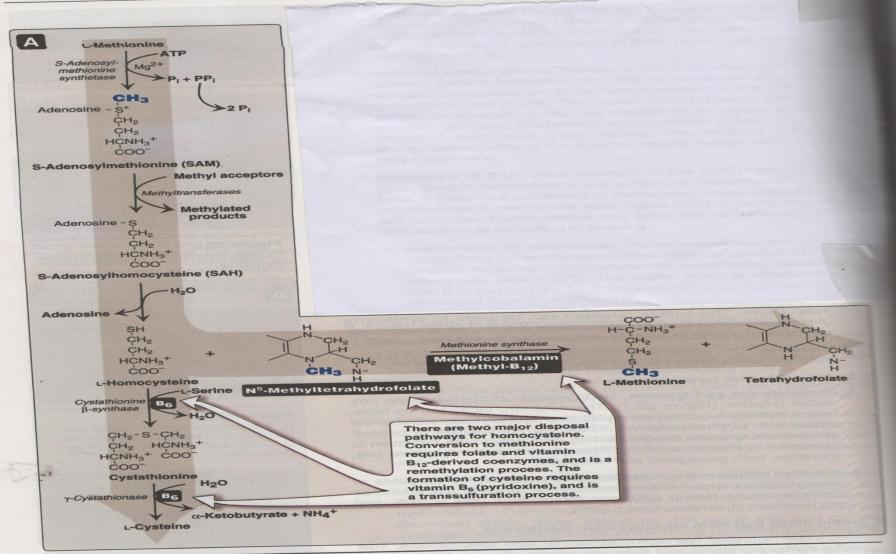
- Methionine is "glucogenic": Propionyl-CoA the endproduct is glucogenic.
- Cysteine formation: (see stage 2)
- "methyl group" and can form choline from ethanolamine. Choline is lipotropic and prevents accumulation of fat in Liver.
- **Polyamine synthesis:** 'Active' methionine after decarboxylation combines with putrescine to form first polyamine **Spermidine** (Refer, biogenic amines).

 Synthesis and degradation of S-adenosylmethionine (SAM).
 (From methionine)

Amino Acid Metabolism

3. Transmethylation

- Methyl group of a methyl donor (e.g. methionine) is <u>transferred</u> to other substances (methyl acceptors) ----- catalyzed by <u>methyl</u> transferases or trans-methylases (such methyl group is known as "labile methyl" group).
- Methionine cannot directly supply its -CH₃ group but it has first to be converted to "active methionine" (SAM).
- Methionine condenses with ATP, forming SAM —--- an unusual high energy compound as it contains no phosphate.



Degradation and resynthesis of methionine. [Note: The resynthesis of methionine from homocysteine is the only reaction in which THF both carries and donates a methyl group. In all other reactions, SAM is the methyl group carrier and donor.]

Transmethylation reactions

S-Adenosyl Methionine Guanido acetic acid ('Active' Methionine) Creatine Ethanolamine (three methylations) Choline **Nicotinamide** N'-methyl nicotinamide Nor-epinephrine ▶ Epinephrine Carnosine (5) Anserine N-acetyl serotonin **(6**) Melatonin Cytosine 5-methyl cytosine

S-adenosyl-homocysteine Other examples are:

- 9. Uracil

 → Thymine
- 10. Orthomethylation of oestrogens.

Elevated homocysteine and decreased folic acid levels in pregnant women are associated with increased incidence of neural tube defects (improper closure, as in spina bifida) in the fetus. Periconceptual supplementation with folate reduces the risk of such defects.

HOMOCYSTINURIA

3. Homocystinuria Type-1 (classical type)

An inborn error of metabolism, which involves the catabolism of methionine or more specifically its metabolic intermediates homocysteine/and homocystine.

Enzyme deficiency: Genetic deficiency of the enzyme **cystathionine synthetase.** The enzyme defect leads to accumulation of homocystine. Plasma level of homocystine increases and excreted in urine ("overflow" aminoaciduria), 50 to 100 mg or more excreted in urine per day. In some cases, Sadenosyl methionine is also excreted.

Incidence: 1 in 60,000 live births.

Clinical features

- Mental retardation: In children and surviving adults.
- Some affected individuals, are extraordinarily tall, with long extremities, frequently with flat feet with toes out (Charlie-Chaplin gait).
- Liver is enlarged (hepatomegaly).
- Skeletal deformities: Involving spine, (vertebrae), and thorax, resulting to kyphosis, scoliosis, arachnodactyly. May

be premature osteoporosis which also accounts to above deformities. X-ray spine shows **cod fish** Vertebrae.

- Ectopia lentis: Curious dislocation of lens of the eye. Not seen at birth, may show at the age of 2 to 3 years.
- Life-threatening arterial/venous thrombosis.
- Most of the patients show abnormal EEG.

Urine: Sodium cyanide-nitroprusside test is positive and helps in diagnosis.

The classical type of homocystinuria is described above. In addition to above classical type, two more types of homocystinurias have been described.

a. Homocystinuria Type-2

- Inheritance: Autosomal recessive
- Enzyme deficiency: N⁵-methyl-Tetrahydrofolatehomocysteine methyl transferase.
- Clinical feature
 - Mental retardation +
 - No ectopia lentis or thrombotic episodes seen.

Blood: Shows increased level of homocysteine.

Urine: Homocysteine is excreted in urine. Nitroprusside test

b. Homocystinuria Type-3

- · Inheritance: Autosomal recessive.
- Enzyme deficiency: N⁵, N¹⁰-methylene tetrahydrofolate reductase deficiency.
- Clinical features
 - Mental retardation +
 - No ectopia lentis or thrombotic episodes

Blood: Shows increase homocysteine.

Urine: Excretion of homocystine, nitroprusside test +ve.

Note: Both type 2 and type 3 show response to folic acid administration.

Metabolism of Cystine

B. Metabolic Role of Cysteine

- Glucogenic: Cysteine is catabolised to Pyruvic acid which is glucogenic.
- Formation of glutathione: Cysteine is required for synthesis of glutathione. G-SH is the reduced form, active group is SH group. G-S-S-G is the oxidised form.
- Formation of taurine: Cysteine is utilised in the formation of 'taurine', which combines with cholic acid (obtained from degradation of cholesterol in Liver,) to form Bile acid 'taurocholic acid'.

CLINICAL ASPECT

INHERITED DISORDERS OF S-CONTAINING AMINO ACIDS

1. Cystinuria

An inherited disorder of cystine metabolism. Excretion of cystine in urine increases 20 to 30 times of normal. Also there occurs increased excretion of diabasic amino acids • lysine, • arginine and • ornithine (specific diabasic aminoaciduria).

Defect: It is considered to be due to a renal transport defect in that reabsorption of the above four amino acids do not occur, a single reabsorptive site is involved.

Complications: Cystine is relatively insoluble amino acid, which may precipitate in renal tubules, ureters and bladder to form cystine calculi. Cystine stones account for 1 to 2 per cent of all urinary tract calculi. It forms a major complication of the disease. A mixed disulphide consisting of L-cysteine and L-homocysteine has been found in urine. This is more soluble and thus reduces the tendency to formation of cystine crystals/ and calculi.

Diagnosis

- brine examination: Detection of hexagonal, flat crystals in urinary deposit in a patient who is not taking sulpha drugs is pathognomonic.
- Cyanide-nitroprusside test (Lewis): It is a simple and valuable test. Urine sample is made alkaline with ammonium hydroxide and then sodium cyanide is added and mixed. Sodium cyanide reduces cystine, if any present, to cysteine. Cysteine forms magenta-red colour, when sodium

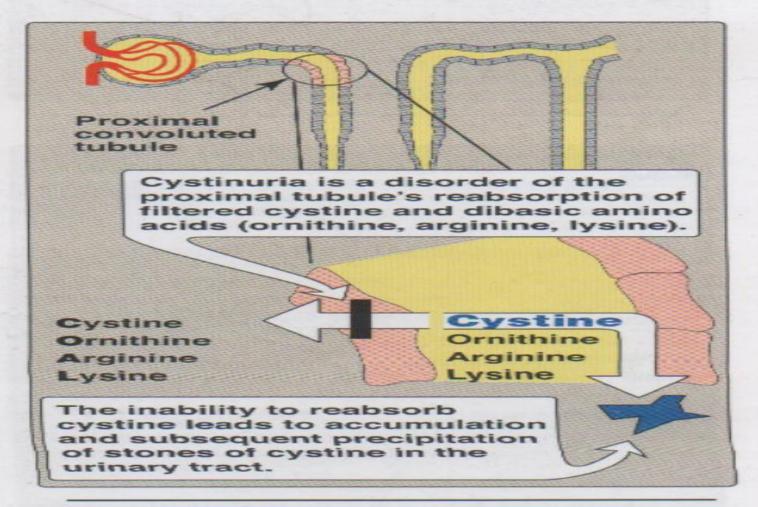


Figure 19.6

Genetic defect seen in cystinuria. [Note: Cystinuria is distinct from cystinosis, a rare defect in the transport of cystine out of lysosomes that results in the formation of cystine crystals within the lysosome and tissue damage.]

Synthesis of Polyamines

Synthesis of Polyamine

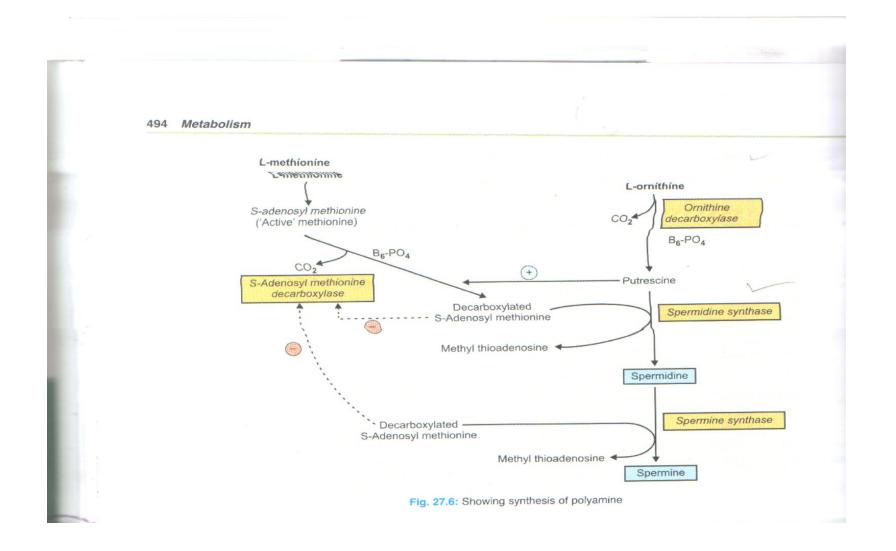
Polyamines

Types of Polyamines are:

- Spermidine
- Spermine

Ornithine in addition to its role in urea cycle, serves the precursor of ubiquitous mammalian and bacterial polyamines, spermidine and spermine. It requires 'active' methionine. Normal human can synthesise approx 0.5 n mol of spermine/day.

Synthesis of Polyamine



Functions of polyamines

- 1. They are growth factors required for the growth and proliferation of cells
- 2. Spermidine has been claimed to be the best marker at tumour cell destruction.

Polyamine Inhibitors

Are used for:

- 1. Treatment of cancer cells
- 2. Treatment of Trypanosomiasis (Sleeping Sickness)
- 3. As hair growth inhibiting agent (Topical applications)